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Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

A SIMPLE PREPARATION OF 1-HYDROXYPYRENE

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To cite this Article Sehgal, Raj K. and Kumar, Subodh(1989) 'A SIMPLE PREPARATION OF 1-HYDROXYPYRENE', *Organic Preparations and Procedures International*, 21: 2, 223 — 225

To link to this Article: DOI: 10.1080/00304948909356367

URL: <http://dx.doi.org/10.1080/00304948909356367>

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General Procedure for the Oxidation of 4-Chromanones with m-CPBA.- 4-Chromanones (1a-e) (0.01 mol) and m-chloroperbenzoic acid (1.72 g, 0.01 mol) were heated under reflux in dry dichloromethane (50 ml) over a period of 15-18 hrs. The m-chlorobenzoic acid which had precipitated during reflux was removed by filtration and the filtrate was concentrated. The resulting residue dissolved in ethyl acetate and the solution was washed with 2% aqueous sodium bicarbonate (3 x 30 ml) and dried over anhydrous sodium sulphate and evaporated. The colorless semi-solids were crystallized from pet. ether (IIb, c) or benzene (IIc, e) to yield colorless needles. IIa was purified by preparative thin layer chromatography to yield a colorless oil.

Synthesis of 3,4-Dihydro-1,5-benzodioxepin-2-one (IIa) from Catechol.- Catechol (1.1 g, 0.01 mol) and β -chloropropionyl chloride (1.2 g, 0.01 mol) was stirred at room temperature in 5% methanolic potassium hydroxide solution (50 ml) for 6 hrs. The solvent was removed under reduced pressure. The residue (1.0 g) was chromatographed over silica gel (ACME, 200 mesh, 60 g). Elution with benzene (200 ml) gave a crude liquid which was purified by preparative TLC, to yield 3,4-dihydro-1,5-benzodioxepin-2-one (IIa) as colourless liquid (0.41 g, 20%), identical in all respects (Co-tlc and Superimposable IR) with IIa prepared as described above.

Acknowledgements.- One of the authors (M. S. N. R.) is grateful to Council of Scientific and Industrial Research, New Delhi, for the award of Junior Research Fellow.

REFERENCES

1. C. Prasadrao and G. Srimannarayana, Syn. Comm. 17, 1507 (1987).
2. K. C. Reddy, B. Veeramallaiah and G. Srimannarayana, Curr. Sci., 49, 18 (1980).
3. F. Eiden and C. Schmiz, Arch. Pharm. (Weinheim), 312, 741 (1979).
4. A. Murata, T. Ito, K. Fujiyasu and T. Suzuki, Bunseki Kagaku, 15, 143 (1966); Chem. Abs., 65, 6275 (1966).
5. P. F. Wiley, J. Am. Chem. Soc., 73, 4205 (1951).

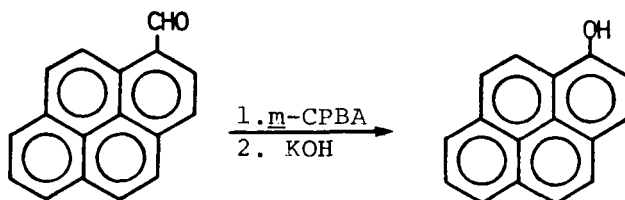
A SIMPLE PREPARATION OF 1-HYDROXYPYRENE

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(04/04/88)

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1-Hydroxypyrene, a metabolite¹ of pyrene, is a valuable intermediate for the synthesis of

various phenolic metabolites of benzo(a)pyrene² and other oxygenated derivatives of pyrene.³ The reported procedures^{4,5} for the synthesis of 1-hydroxypyrene are unsatisfactory both because of the very poor yields of 1-hydroxypyrene and the difficulty in its preparation, especially on a large scale (e. g. 50 g). We have recently developed a relatively simple and high yield procedure for the synthesis of 1-hydroxypyrene from pyrene-1-carboxaldehyde using a method analogous to the one originally developed by Godfrey *et al.*⁶ for the synthesis of methoxy phenols. Thus, Baeyer-Villiger oxidation of pyrene-1-carboxaldehyde with an excess



of *m*-chloroperoxybenzoic acid in boiling dichloromethane followed by hydrolysis of the resultant formate with alkali produced the desired 1-hydroxypyrene in a 90-96% yield.

EXPERIMENTAL SECTION

1-Hydroxypyrene.- Pyrene-1-carboxaldehyde (46 g, 0.2 mol) and *m*-chloroperoxybenzoic acid (52 g, 0.3 mol) were heated under reflux in dry dichloromethane (800 mL) for 24 hrs under an argon atmosphere. Most of the solvent was removed by distillation under reduced pressure, and the residue was stirred with 10% NaHCO₃ (500 mL) until effervescence ceased. The residual dichloromethane was evaporated by using water aspirator, and the resulting crystalline solid was collected, washed with water and dried. The air dried solid was dissolved in a mixture of THF (200 mL) and methanol (200 mL), and then 60 mL of 25% KOH solution was added. The mixture was stirred under argon for 3-4 hrs, most of the solvent was removed by distillation under reduced pressure, and the residue was then diluted with 1 l. of 2% KOH. The unreacted aldehyde and other neutral impurities were extracted with benzene-ether (1:1), and the ice-cooled aqueous phase was acidified to pH 2. The resultant solid was collected, washed with water and then triturated several times with 10% NaHCO₃. The gray solid was finally washed with water and dried *in vacuo* to yield 42.0 g (92%) of 1-hydroxypyrene, mp. 176-178°. A small sample was further purified according to the procedure described earlier⁵ to give a pale yellow crystalline solid, mp. 178-180°, lit.⁵ mp. 179-180° with a little loss.

Acknowledgement.- This work was supported in part by Grant ESO4130 awarded to Subodh Kumar by the National Institute of Environmental Health Sciences, DHHS.

REFERENCES

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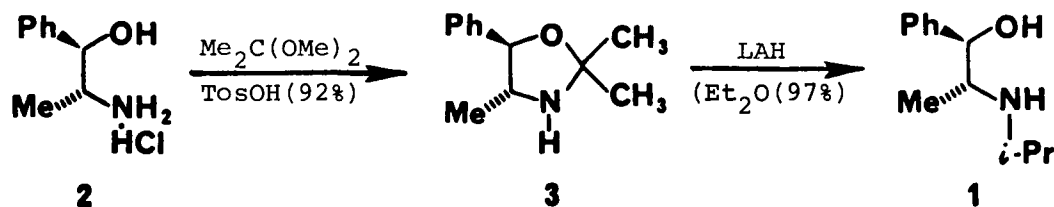
1. P. Sims, *Biochem. Pharmacol.*, **19**, 795 (1970); E. Boyland and P. Sims, *Biochem. J.*, **90**, 391 (1964).
2. A. Graslund, F. Waern and B. Jernstrom, *Carcinogenesis*, **7**, 167 (1986); O. Ribeiro, C. A. Kirkby, P. C. Hirom and F. Oesch, *ibid.*, **8**, 1621 (1987).
3. E. Profit and R. Biela, *Chem. Ber.*, **94**, 2374 (1961).
4. H. Vollmann, H. Becker, M. Corell and H. Streeck, *Ann.* **531**, 1 (1937); E. Tietze and O. Bayer, *ibid.*, **540**, 189 (1939).
5. W. H. Gumprecht, *Org. Syn. Coll. Vol. 5*, 632 (1973).
6. I. M. Godfrey, M. V. Sargent and J. A. Fix, *J. Chem. Soc. Perkin Trans. I*, 1353 (1974).

AN IMPROVED SYNTHESIS OF

(1R, 2R)-(-)-a-(1-ISOPROPYLAMINOETHYL)BENZYL ALCOHOL

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Directed asymmetric synthesis via 1,4-addition reactions of chiral allylphosphonyl anions with cyclic enones is a powerful new synthetic tool.¹ The chiral director (1R, 2R)-(-)-a-(1-isopropyl-aminoethyl)benzyl alcohol (1) was originally prepared from the coupling reaction of (1R, 2R)-(-)-norpseudoephedrine hydrochloride (2) with acetone-sodium acetate-sodium borohydride in acetic acid-water at 0° followed by reduction of the resulting (4R, 5R)-(-)-5-phenyl-2,2,4-trimethyl-1,3-oxazolidine (3) with lithium aluminum hydride in ether at room temperature.^{1,2} We later found that in the first step, the formation of the intermediate, oxazolidine 3 is very much dependent on the sodium borohydride used. The reaction proceeded to varying degree depending on the lot number of sodium borohydride employed. These undependable results prompted us to explore a different method for the formation of 3. Herein, we now report an improved and simple procedure for the preparation of 1 in excellent yield.



Treatment of amino alcohol 2 with 2,2-dimethoxypropane and anhydrous *p*-toluene